

Agnes Pockels: Life, Letters and Papers

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Agnes Wilhelmine Louise Pockels

14.2.1862 - 21.11.1935



Born in Venice to parents originally from Northern Germany

Father: officer in the Austrian Army.

One brother, Fritz, born 1865 in Vincenca.

Problem: malaria wide-spread in North-Italy.

⇒ retirement of Captain Pockels in 1871



Theodor Pockels, father (1830-1906)



1872-1877: Municipal High School for Girls, Brunswick





No degree available for girls, two natural science classes per week in the final two years of high school

- " (...) without conferring to a special scientific systematic thought" (curriculum)
- "I had a passionate interest in natural science, especially physics, and would have liked to study" (Agnes Pockels) in Autobiographical Notes in W. Ostwald (1932), Kolloid-Zeitschrift, 58, 1
- ⇒ retreat to private life after school (care for sickly parents)

1883: Brother Fritz begins his studies in physics



Fritz Pockels



- ⇒ access to physics and mathematics books and autodidactic learning
- ⇒ access to "Naturwissenschaftliche Rundschau" (akin to Physics Today)

Agnes and Fritz Pockels: Working at Home







Agnes Pockels

"Their studies were often conducted in the kitchen where Agnes, being the daughter of the household, hat to do her chores and where it was always nice and warm. In this way, Agnes conducted her first observations in the field of surface science. This is really true, and no joke or poetic licence: what millions of women see every day without pleasure and are anxious to get rid of, i.e. the greasy washing-up water, encouraged this girl to make observations and eventually do ... scientific investigations."

Electro-Optic **Effect**

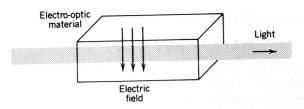


Figure 18.0-1 A steady electric field applied to an electro-optic material changes its refractive index. This, in turn, changes the effect of the material on light traveling through it. The electric field therefore controls the light.



Friedrich Carl Alwin Pockels (1865 - 1913)

Ph.D. from Göttingen University in 1888

1900 - 1913 Prof. of theoretical physics in Heidelberg

for certain materials n is a function of E, as the variation is only slightly we can Taylor-expand n(E):

$$n(E) = n + a_1 E + \frac{1}{2} a_2 E^2 + \dots$$

linear electro-optic effect (Pockels effect, 1893):

$$n(E) = n - \frac{1}{2}r \cdot n^3 E \qquad r = -2\frac{a_1}{n^3}$$

quadratic electro-optic effect (**Kerr effect**, 1875):

$$n(E) = n - \frac{1}{2} s \cdot n^3 E^2$$
 $s = -\frac{a_2}{n^3}$

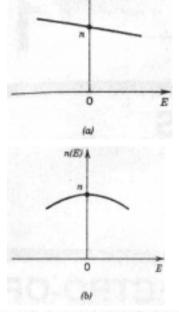
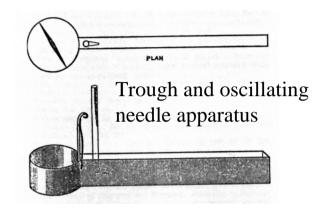


Figure 18.1-1 Dependence of the refractive index on the electric field: (a) Pockels medium; (h) Kerr medium.

Rayleigh's Interest in Water Surfaces



Rayleigh, Lord, Proc.

R. Soc. London 1890,47,281

Idem, ibid. 1890,47,364

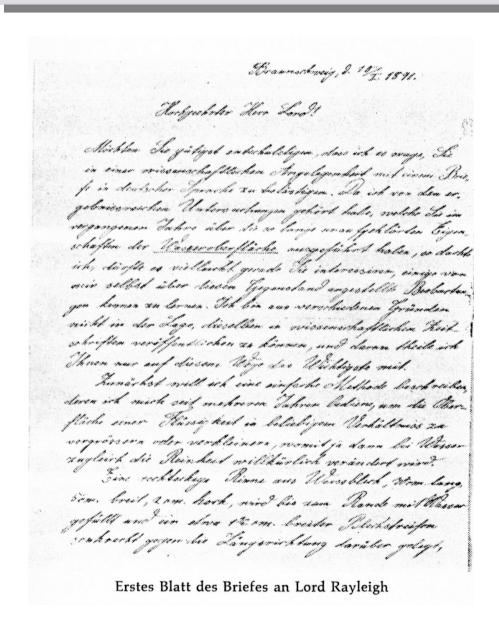
Idem, ibid. 1890,48,127

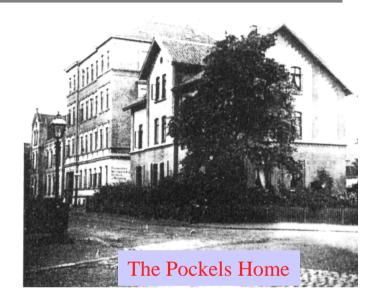


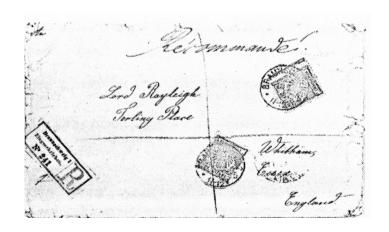
John William Strutt, 1842-1919 Third Baron Rayleigh (since 1873)

- Surface tension decreased if a body is inserted into the liquid (soap or camphor) or by ,contamination' with a surface film of insoluble grease or oil (or finger in water)
- Wave damping by a drop of oil placed on the surface (waves produced by a fan)
- Powder (sulphur) disperses uniformly on <u>clean</u> water (Aitken)
- ? films of oil or grease on water ultimately extend until they become one molecule thin
- ? a method to determine the exact degree to which the oil film must be extended to form a complete single layer of molecules, each in contact with its nearest neighbours
- ? precise determination of the surface tension of oily surfaces

Agnes Pockels: A Letter to Rayleigh after Reading about his Results in "Naturwissenschaftliche Rundschau"









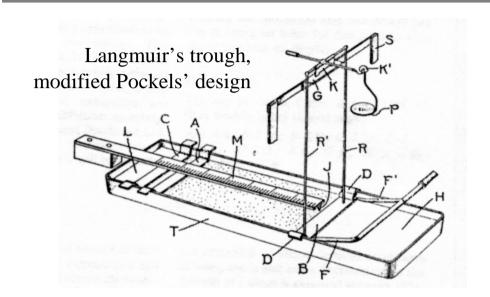
Agnes Pockels

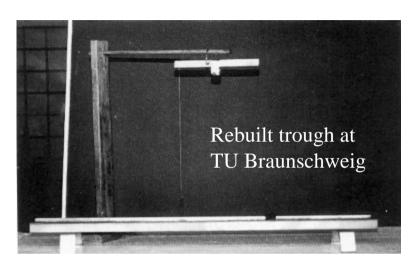
"My lord, will you kindly excuse my venturing to trouble you with a German letter on a scientific subject? Having heard of the fruitful researches carried on by you last year on the hitherto little-understood properties of water surfaces, I thought it might interest you to know of my own observations on the subject. For various reasons I am not in a position to publish them in scientific periodicals, and I therefore adopt this means of communicating to you the most important of them.

I thought I ought not to withhold from you these facts which I have observed, although I am no professional physicist; and again begging you to excuse my boldness, I remain with sincere respect."---POCKELS, January 10

"I shall be obliged if you can find space for the accompanying translation of an interesting letter which I have received from a German lady, who with very homely appliances has arrived at valuable results respecting the behaviour of contaminated water surfaces. The earlier part of Miss Pockels' letter covers nearly the same ground as some of my own recent work, and in the main harmonizes with it. The later sections seem to me very suggestive, raising, if they do not fully answer, many important questions. I hope soon to find opportunity for repeating some of Miss Pockels' experiments." ---- RAYLEIGH, March 2

Agnes Pockels: Tools for Characterizing the Water Surface

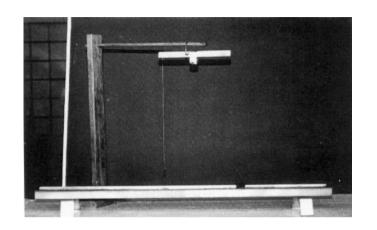




- Rectangular tin trough, 70 cm long, 5 cm wide and 2 cm high filled with water to the brim and a strip of tin about 1.5 cm wide laid across it perpendicular to its length, so that the underside of the strip is in contact with the water.
- Variation of surface area by altering the strip position.
- Surface tension measured with an apothecary's balance to determine the weight necessary to lift a small disk, 6 mm diameter (i.e. a button) from the water.

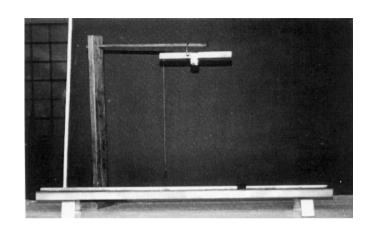
She sent her results to the professors of physics of the University of Göttingen who did not express any special interest in them.

Agnes Pockels, **Surface Tension** (1891) *Nature* **46**, 437: Ten Years of Research in One Paper



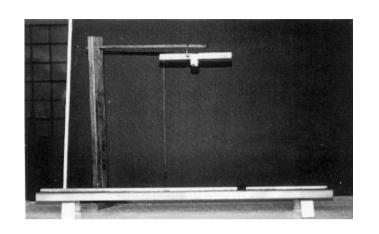
- I. The surface tension of a ,contaminated water surface varies if the surface is compressed or expanded. On compression, the surface tension decreases up to a factor of two. On expansion, it increases until a maximum value is reached, then remains constant.
- II. Thus, a water surface can exist in two states, a *normal* condition in which the surface tension remains constant if the surface size is changed, and an *anomalous* condition, where varying the surface size leads to a change in surface tension.
- III. The cleaner the surface, the more it can be compressed while the surface pressure remains at its maximum value.
- IV. A surface in the *anomalous* state can be cleaned by removing the surface layer by immersing and withdrawing a paper strip.
- V. The lower the surface tension, the more pronounced is the wave damping.

Agnes Pockels, **Surface Tension** (1891) *Nature* **46**, 437, cont.: Ten Years of Research in One Paper



- VI. All solid bodies, no matter how clean, contaminate a water surface that is in *normal* state.
- VII. Solid materials like glass or metals increase the relative ,contamination'.
- VIII. Other materials like camphor or flour reduce the surface tension until an equilibrium value is reached that is different for each substance. The solid is the source of a current of ,contaminating material. A **contact line** between water surface and the **material current** can be **made visible** by dusting the water with flowers of sulfur.
- IX. A material current occurs between surfaces with the same surface tension if the ,contamination' is caused by different substances. "Equal relative contamination by different substances does not indicate equality of **that (osmotic?) pressure -in surface films -** which is the cause of the material current between surfaces of equal surface tension."

Agnes Pockels, **Surface Tension** (1891) *Nature* **46**, 437, cont.: Ten Years of Research in One Paper



X. Solutions of sugar have a *normal* surface and a higher surface tension than pure water. Yet a piece of sugar brought to its surface causes material current and surface tension reduction.

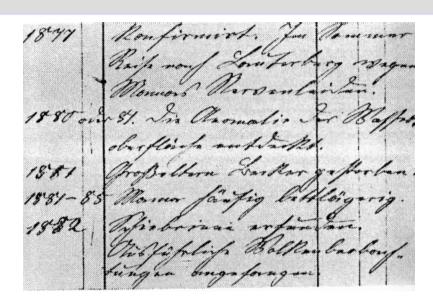
XI. "From these experiments I concluded (a) that the **surface layer of water can take up** *more* **of soluble substances than the bulk solution**; (b) that the compressed surface of a solution gets *anomalous*, always and only, if it contains more of the dissolved substance than the bulk."

Basis for Quantitative Study of Surface Films



The Pockels Domestic Scene

Diary until 1907

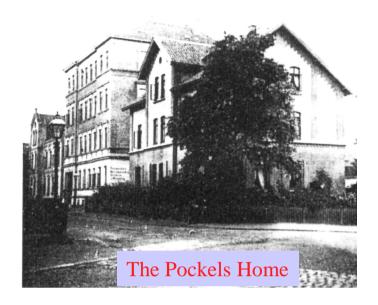


'1877: ... Mummy's nerve trouble;' '1881-85: Mummy often ill in bed;' '1887: In winter, Daddy had carbuncles;' '1888: In summer, Daddy had catarrh of the bladder;' '1890: Daddy was ill in March and October;' '1892: In January, Fritz had influenza. In March and April, Daddy was ill...At the end of July... Daddy was ill;' '1894: In February, Fritz had influenza. At Easter ... Daddy was ill;' ...

Elisabeth Pockels, Bericht der Oberhessischen Gesellschaft für Natur-und Heilkunde 24 (1949) 303 "the family suffered ... with much illness, and some pessimism was also in evidence so that a large part of her life, during the years 1890-1914, was spent in home nursing. Her scientific work had to take second place... When she was warned to pay attention to her own health, she said once: "Like a soldier, I stand firm at my post caring for my aged parents."

Agnes Pockels: Her Life

Autobiographical Notes in W. Ostwald (1932), *Kolloid-Zeitschrift*, 58, 1



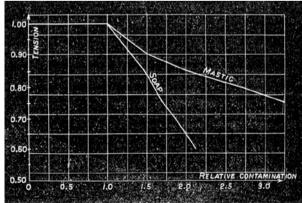
"Encouraged by the support of Lord Rayleigh and the publications in *Nature,* I dared to submit papers to German journals, and conducted about 10 years of successful investigations. The care of my sick parents was very challenging, therefore after 1902 I did few experiments, but I did a translation and wrote comments for "Beiblätter zu den Annalen", which my brother edited. In 1909 I published my observations of the contact angle."

Agnes Pockels: Further Publications (chronological)

- (1) **Surface Tension**, (1891) *Natur*, **46**, 437.
- (2) On the relative contamination of the water surface by equal quantities of different substances, (1892) *Nature* 47, 418.
- (3) Relations between the surface tension and relative contamination of water surfaces, (1893) *Nature*, **48**, 152.
- (4) On the spreading of oil upon water, (1894) *Nature* **50**, 223.
- (5) **Beobachtungen über die Adhäsion verschiedener Flüssigkeiten an Glas**, (Observations about the Adhesion of Different Liquids on Glass), (1898) *Naturwissenschaftliche Rundschau*, **14**, 190.
- (6) Randwinkel gesättigter Lösungen an Kristallen (Contact Angles of Saturated Solutions on Crystals), (1899), *Naturwissenschaftliche Rundschau*, **14**, 383.
- (7) Untersuchungen von Grenzflächenspannungen mit der Cohäsionswaage, (Investigations of the Surface Tension with the Cohesion Balance), (1899) *Annalen der Physik*, **67**, 668.
- (8) Über das spontane Sinken der Oberflächenspannung von Wasser, wässerigen Lösungen und Emulsionen, (On the Spontaneous Decrease of the Surface Tension of Water, Aqueous Solutions and Emulsions), (1902) Annalen der Physik, **8**, 854.
- (9) Über Randwinkel und Ausbreitung von Flüssigkeiten auf festen Körpern (On Contact Angles and the Flow of Fluids on Solid Bodies), (1914) *Physikalische Zeitschrift*, **15**, 39.
- (10) Zur Frage der zeitlichen Veränderung der Oberflächenspannung (On the Changes of the Surface Tension with Time), (1916) *Physikalische Zeitschrift*, **17**, 141
- (11) Über die Ausbreitung reiner und gemischter Flüssigkeiten auf Wasser (On the Spreading of Pure and Mixed Liquids on Water) (1916) *Physikalische Zeitschrift*, **17**, 142.
- (12) **Die Anomalie der Wasseroberfläche** (The Anomalous State of the Water Surface) (1917) *Die Naturwissenschaften*, **5**, 137 u. 149.
- (13) Zur Frage der Ölflecke auf Seen (On Oil Stains on Lakes) (1918) Die Naturwissenschaften, 6, 118.
- (14) The measurement of surface tension with the balance (1926) Science 64, 304.

Agnes Pockels: The Subsequent Nature Papers

- I. "In order to divide very small masses exactly I chose the course of Lord Rayleigh of transferring the contaminating substance to the water-surface by means of a volatile solvent. But instead of ether I used *benzine*, and **let the drops of the solution evaporate directly from the water** instead of vaporizing them on a metal plate and then immersing this, as Lord Rayleigh did."
- II. The area of onset of a decrease in surface tension: 1mg provencer oil 8460cm², 1mg stearic acid 4711cm², etc. The thickness of the film is very small, i.e. 1.3nm for provencer oil.
- III. The smallest thickness found amounts to 0.037nm, "It seems to me very probable that the contaminating substances *are not spread in coherent films at all*, but rather in a state of very fine distribution between the superficial water molecules **that must be named either emulsion or solution.**"
- IV. The first isotherms with quantitative units.



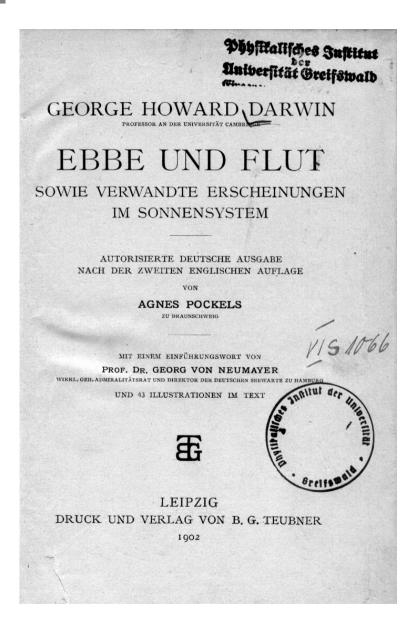
Agnes Pockels: Subsequent Papers

- I. Glass cleaning by heating it in a flame, and immersing it in water. If the glass is clean, no material current should occur.
- II. The material of the trough and the barrier should exhibit very high contact angles with water.
- III. Certain oils yield very homogeneous fluid layers of constant thickness (uniform interference colour); they thin on area expansion until they break at a thickness typical for the oil used.
- IV. Optimizing the ring method to determine surface tension (Science, 1926).

Other Interests: Translation of a Textbook

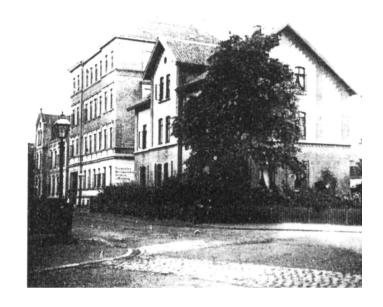
George Howard Darwin

The Tides and Kindred Phenomena in the Solar System



Agnes Pockels: Her Life

Autobiographical Notes in W. Ostwald (1932), *Kolloid-Zeitschrift,* 58, 1



"When my brother died in 1913, the alarums of the war and post-war period engulfed me and the *Beiblätter* (*zu den Annalen der Physik*) ceased pulication; I was no longer in a position to obtain the relevant literature and in the end I completely lost contact with research in my field, the deterioration in my eyesight and my health altogether being a contributing factor."



"During her later years she led a quiet life as 'Auntie Agnes', like many other middleaged women in Braunschweig. She had a wide circle of acquaintances, and two puzzle-solving societies met in her home. The war and the inflation hit this society hard. ...

She herself suffered relatively less because her American relatives provided for her financially, and because later on the capital invested in California produced extremely high dividends. As a result, she was able to help others in all kinds of ways, and shared her income for years. She herself always lived simply, and kept her thoughts to herself without saying much.

The information about her special scientific knowledge was now only mentioned abroad in whispers."

Agnes Pockels: Late Recognition

Annual Laura-R.-Leonard Prize of the German Colloid Society

"for her quantitative investigation of the properties of interfaces and surface films, and for the methods she used, which have since become fundamental in modern colloid science"

She taught us not only cleanness in our work, she also taught us how to measure it! (W. Ostwald (1932), Kolloid-Zeitschrift, 58, 1)

1923	Prof. Dr. Wolfgang Pauli, sen. (1869 - 1955)	Physician (Wien, Au)
1923	Prof. Dr. Richard Zsigmondy (1865 - 1929)	Chemist (Göttingen) Nobel Prize 1925
1924	Prof. Dr. Martin H. Fischer (1879 - 1962)	Physician (Cincinnatti, OH)
1925	Prof. Dr. Henry Siedentopf (1872 - 1940)	Physicist (Jena)
1926	Prof. Dr. Hermann Ambronn (1856 - 1927)	Biologist (Jena)
1927	Prof. Dr. Alfred Lottermoser (1870 - 1945)	Chemist (Dresden)
1928	Prof. Dr. Herbert Freundlich (1880 - 1941)	Chemist (Berlin)
1928	Prof. Dr. Sir William Hardy (1864 - 1934)	Chemist (Cambridge, GB)
1929	Dr. Dr. h. c. R. E. Liesegang (1869 - 1947)	Photographic Factory (Frankfurt/Main)
1930	Prof. Dr. Heinrich Bechhold (1866 - 1937)	Physician (Frankfurt/Main)
1931	Agnes Pockels (1862 - 1935)	Housewife (Braunschweig)
1932	Prof. Dr. Peter P. von Weimarn (1879 - 1935)	Chemist (Kobe, Japan)
1933	••••	

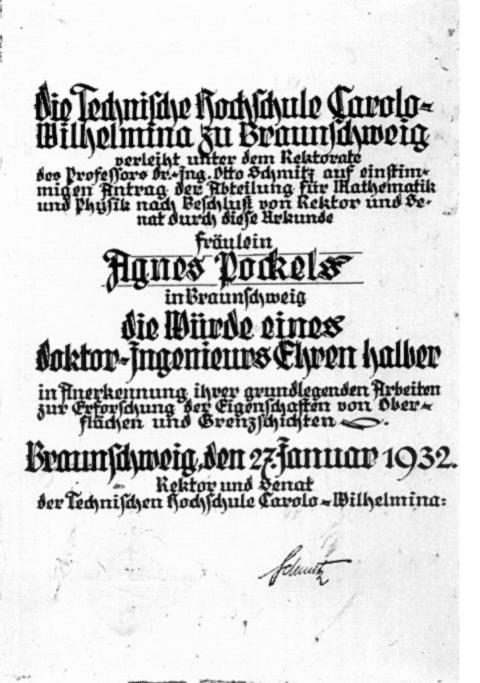
Late Recognition, cont.

1932 First Woman to receive the Dr. h. c. from the Technical University Braunschweig, on the occasion of her 70th birthday

Sonntag, 14. Februar 1932

Die Cacolo: Wilhelmina hat eine Ehrens doktorin ernannt

Seute, am 11. Gebruar, pollendet Graulein Manes Podels ihr 70. Lebensiahr. 211s itille Pripatgelehrte int fie in ihret beimatitadt menig hervorgefreten. Aber im Reiche ber Willenichaft hat fie fich ale initinttfichere Ginberin an bie Seite ber brei Braunidmeiger Großen: Eliter. Gei. tel und Gielel ju itellen vermocht. Die Arbeiten von Graulein Podels, die auf bem Gebiete ber Cheriladenipannungen anienten, erregten bereits por viergig Sahren die Mufmerflam. teit bes berühmten englischen Physiters Lord Ranleigh, ber fich damals mit abnlichen Broblemen beidäftigte. In neuerer Beit ift auf ber Crund. lage ber Podelsiden Arbeiten ein ausgedehnter 3meig ber Rolloidwillenichoit entitanben, Die Lebre pon ben Grensichichten und Gilmen. Aus Diefem Grunde hat die Rolloidgesellichaft an Fraulein Bodels ben "Leonardoreis" für 1931 perlichen und bie Tedniiche Dochicule Braunidmeig hat ihr die Burde eines Dr. . 3 ng. e . h. übertragen. Beibe Chrungen geicahen am Bortage ihres Geburtstages auf einer gerade gufallig in Braunidweig stattfindenden Tagung der Phonitalifden Gefellichaft. Der aus Leipzig gu Diciem 3med ericienene Profesior 2B. Oftwald überreichte die Medaille bes Leonard. preifes und ber Biettor ber Tedniiden Sodicule das Ehrendoftordiplom.



"I learned to my great joy that my work is being used by others for their investigations." (Autobiographical Notes in W. Ostwald (1932), *Kolloid-Zeitschrift*, 58, 1)





THANK YOU!

Heike Schäffner

Uni Greifswald

Elisabeth Grossmann

TU Braunschweig

Sabine Gross

Univ. of Madison, WI

C: State Mecklenburg-Vorpommern

Literature

- (1) Agnes Pockels (ca 1880-1910) **Tagebuch: Lebensereignisse**; Original zusammen mit Familienfotos im Archiv der TU Braunschweig
- (2) Chemical Society (London): Xerographick copies of Agnes Pockels' diary, of the letters she and her brother wrote to Lord Rayleigh (the latter with translations into English) and reprints of photographs of various members of the Pockels family
- (3) Wolfgang Ostwald (1932) **Die Arbeiten von Agnes Pockels über Grenzflächen und Filme**; *Kolloid-Zeitschrift* (58) 1
- (4) Elisabeth Pockels (1949) **Ein gelehrtes Geschwisterpaar. Zur Erinnerung an Agnes Pockels (1862-1935)**; *Bericht der Oberhessenschen Gesellschaft für Natur- und Heilkunde* (24) 303
- (5) Stanley D. Forrester and Charles H. Giles (1971) **The origins of the surface film balance**; *Chemistry and Industry*, 43
- (6) Charles H. Giles und Stanley D. Forrester (1978) **Agnes Pockels und die Ursprünge der Oberflächenchemie** in "Mitteilungen der Technischen Universität Carolo-Wilhelmina zu Braunschweig 13", 15-28.
- (7) Stanley D. Forrester and Charles H. Giles (1979) **Rayleigh and the great monolayer discovery**; *Chemistry and Industry*, 469
- (8) Elizabeth M. Derrick (1982): **Agnes Pockels, 1862-1935**; *Journal of Chemical Education* (59) 1030
- (9) Gabriele Beisswanger (1991): **Agnes Pockels (1862-1935) und die Oberflächenchemie**; *Chemie in unserer Zeit* (25) 97
- (10) Klaus Beneke (1995): "Die Untersuchungen von Agnes Pockels" in "Zur Geschichte der Grenzflächenerscheinungen mit ausgesuchten Beispielen. Mitteilungen der Kolloid-Gesellschaft. Beiträge zur Geschichte der Kolloidgesellschaft, IV ". Verlag Reinhard Knof, Kiel: 40-43
- (11) Liane Biermann: **Agnes Pockels, 1862-1935**; Broschüre der Presse- und Öffentlichkeitsarbeit der TU Braunschweig
- (12) Klaus Benecke (1996) **Laura-R.-Leonard-Preis** in "Über 70 Jahre Kolloid-Gesellschaft, Gründung, Geschichte, Tagungen (mit ausgesuchten Beispielen der Kolloidwissenschaften) Beiträge zur Geschichte der Kolloidgesellschaft, IV ". Verlag Reinhard Knof, Kiel: 94-98